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Basic Atomic Structure
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Worksheet ANSWERS 1

a) protons b) neutrons

c) electrons a) Positive

b) Neutral c) negative

2 atomic number (or
identity); charge 3

protons; electrons (in a
neutral charge atom

only!); same 4 average

atomic weight; mass 5

mass number; nucleus

6 neutrons; protons (or

atomic number); mass

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Structure
number 7 Lithium = Li

Worksheet
= 3 Iron = Fe = 26

Answer Key
Oxygen = O = 8

Krypton = Kr = 36 8 (A
neutral atom by
definition has NO
charge!

Basic Atomic Structure Worksheet ANSWERS

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Worksheet H and the 1.
2. 4. 5. 6. 8. 9. The 3
particles of the atom
are: protons (+) (e- Their
respective charges are:

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The number of protons in one atom of an element determines the atom's number. Of electrons determines the of the element in one atom of an element. It also in a neutral atom of that element. The atomic number

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a) protons b) neutrons
c) electrons a) Positive
b) Neutral c) negative
2 atomic number (or
identity); charge 3
protons; electrons (in a
neutral charge atom
only!); same 4 average
atomic weight; mass 5
mass number; nucleus
6 neutrons; protons (or
atomic number); mass
number 7 Lithium = Li
= 3 Bromine = Br = 35
Iron = Fe = 26 Copper =
Cu = 29 Oxygen = O =
8 Mercury = Hg = 80

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Krypton = Kr = 36

Helium = He = 2 8 (A neutral atom by definition has NO charge!

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Basic Atomic Structure

Worksheet 1. The 3
particles of the atom
are: a. _____ b. _____ c.

_____ Their respective
charges are: a. _____ b.

_____ c. _____ 2. The
number of protons in
one atom of an

element determines
the atom's _____, and
the number of
electrons determines
the _____ of the

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Worksheet

Basic Atomic Structure Worksheet

Write two algebraic equations relating the number of protons and neutrons in an atom to that atom's atomic mass and atomic number: Where, P = Number of protons. N = Number of neutrons. A = Atomic mass. Z = Atomic number

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Science Grade/level: 8

Age: 12+ Main content:

Atoms Other contents:

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protons, neutrons,

electrons, nucleus Add

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number of protons in one atom of an element determines the atom's number of electrons determines the of the element. The atomic number tells you the number of in one atom of an element. It also tells you the number of

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Atomic Structure Worksheet. Define mass number: The

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mass number is an integer (whole number) equal to the sum (addition) of the number of protons and neutrons in an atomic nucleus. Define atomic number: The number of protons in the nucleus of an atom. The three subatomic particles of the atom are: Protons, electrons, neutrons

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defined as the study of
matter in terms of its
structure, composition
and the properties. As
you are aware, matter
is made up of atoms,
and therefore an
understanding of the

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